

N-Phenylindazole-1-thiocarboxamide

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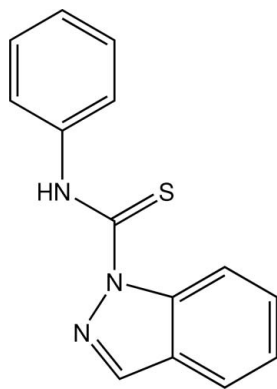
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Key indicators: single-crystal X-ray study; $T = 98$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.030; wR factor = 0.069; data-to-parameter ratio = 13.5.

The title compound, $\text{C}_{14}\text{H}_{11}\text{N}_3\text{S}$, exists as a thione and has a Z conformation about the central $\text{C}-\text{N}$ bond. The phenyl ring is twisted out of the plane defining the rest of the molecule [dihedral angle = 64.73 (8°)]. In the crystal structure, molecules associate into pairs *via* $\text{N}-\text{H}\cdots\text{N}$ bonds and stack into columns *via* $\text{C}-\text{H}\cdots\pi$ interactions.

Related literature

For related literature, see: Hossaini Sadr, Sardroodi *et al.* (2005); Hossaini Sadr, Jalili *et al.* (2005); Henderson *et al.* (2006).



Experimental

Crystal data

$\text{C}_{14}\text{H}_{11}\text{N}_3\text{S}$	$V = 1188.0$ (8) Å ³
$M_r = 253.32$	$Z = 4$
Monoclinic, $C2$	Mo $K\alpha$ radiation
$a = 20.408$ (8) Å	$\mu = 0.26$ mm ⁻¹
$b = 5.690$ (2) Å	$T = 98$ (2) K
$c = 11.949$ (5) Å	$0.42 \times 0.12 \times 0.02$ mm
$\beta = 121.110$ (14)°	

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Data collection

Rigaku AFC12κ/SATURN724 diffractometer	6334 measured reflections
Absorption correction: multi-scan (<i>ABSCOR</i> ; Higashi, 1995)	2202 independent reflections
$T_{\min} = 0.506$, $T_{\max} = 1.000$	2160 reflections with $I > 2\sigma(I)$
(expected range = 0.503–0.995)	$R_{\text{int}} = 0.023$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.030$	H-atom parameters constrained
$wR(F^2) = 0.069$	$\Delta\rho_{\text{max}} = 0.22$ e Å ⁻³
$S = 1.05$	$\Delta\rho_{\text{min}} = -0.19$ e Å ⁻³
2202 reflections	Absolute structure: Flack (1983),
163 parameters	846 Friedel pairs
1 restraint	Flack parameter: 0.03 (7)

Table 1

Hydrogen-bond geometry (Å, °).

C_g is the centroid of the C_9-C_{14} ring.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N}3-\text{H}3\cdots\text{N}2^{\text{i}}$	0.88	2.30	3.097 (3)	151
$\text{C}5-\text{H}5\cdots\text{S}1^{\text{ii}}$	0.95	3.14	3.881 (2)	136
$\text{C}6-\text{H}6\cdots\text{S}1^{\text{iii}}$	0.95	3.17	4.040 (3)	153
$\text{C}1-\text{H}1\cdots\text{C}_g^{\text{iv}}$	0.95	2.47	3.389 (3)	164

Symmetry codes: (i) $-x + 1, y, -z$; (ii) $-x + 1, y + 1, -z + 1$; (iii) $-x + 1, y, -z + 1$; (iv) $-x + 1, y + 1, -z$.

Data collection: *CrystalClear* (Rigaku Americas Corporation, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BV2071).

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supplementary materials

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***N*-Phenylindazole-1-thiocarboxamide**

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Comment

The preparation and crystal structures of series of compounds containing molecules related to the title compound (I), *i.e.* with the general formula PhNC(=S)R ($R = \text{pyrazolyl, 3,5-dimethylpyrazolyl and 3,5-diphenylpyrazolyl}$) and $\text{PhNC(=S)NRR}'$ ($R = \text{alkyl or aryl}$) have been reported recently (Hossaini Sadr, Sardroodi *et al.*, 2005; Hossaini Sadr, Jalili *et al.*, 2005; Henderson *et al.*, 2006). Compound (I) exists as a thione, C8=S1 is 1.6545 (19) Å, and the conformation about the C8—N3 bond is *Z*. The molecule is non-planar as the phenyl ring is twisted out of the plane of the central chromophore as manifested in the C8/N3/C9/C10 torsion angle of 51.7 (2)°. In the crystal structure, molecules related by 2-fold symmetry associate *via* $\text{N—H}\cdots\text{N}$ hydrogen bonds, Table 1. Pairs of molecules are connected by $\text{C—H}\cdots\pi$ interactions [$\text{C1—H1}\cdots\text{Cg}(\text{C9—C14}) = 2.47$ Å for symmetry operation $1 - x, 1 + y, -z$] on either side to form columns parallel to the *b* axis. Connections between columns are afforded by weak $\text{C—H}\cdots\text{S}$ interactions; the S1 forms two such contacts with different molecules of the adjacent chain, Table 1. A view of the unit-cell contents is shown in Fig. 2.

Experimental

A solution of indazole (1.18 g, 10 mmol) in dry *n*-hexane (25 ml) was treated with solid NaH (55%, 0.44 g, 10 mmol) under an N_2 atmosphere. Liberation of H_2 stopped after 3 h stirring. PhNCS (1.2 ml, 10 mmol) was added to the resulting suspension of $[\text{Na}][\text{Indz}]$ and the reaction was continued overnight at r.t. The suspension was filtered using a fritted funnel and the collected white solid, $[\text{Na}][\text{PhNCSIndz}]$, was washed with cold *n*-hexane (3 x 10 ml) and dried *in vacuo*; yield 2.60 g, 95%. *M.p.* = 240 °C (melting along with colour change and decomposition). Colourless crystals of (I) were obtained unintentionally as the decomposition product of an authenticated sample of $\text{Bu}_2\text{Sn}(\text{PhNCSIndz})_2$ during attempts to grow crystals from heptane solution in air; *M.p.* = 90 °C.

Refinement

All C- and N-bound H atoms were included in the riding-model approximation, with $\text{C—H} = 0.95$ Å and $\text{N—H} = 0.88$ Å, and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C, N})$.

Figures

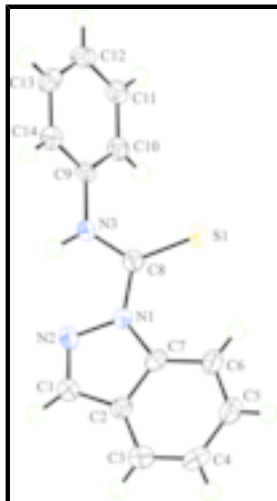


Fig. 1. Molecular structure of (I) showing atom-labelling scheme and displacement ellipsoids at the 70% probability level.



Fig. 2. View of the crystal packing in (I) down the *b* axis. The N—H...N hydrogen bonds are shown as orange-dashed lines. Colour code: yellow (sulfur), red (oxygen), blue (nitrogen), grey (carbon) and green (hydrogen).

***N*-Phenylindazole-1-thiocarboxamide**

Crystal data

$C_{14}H_{11}N_3S$

$M_r = 253.32$

Monoclinic, *C*2

Hall symbol: *C* 2y

$a = 20.408 (8) \text{ \AA}$

$b = 5.690 (2) \text{ \AA}$

$c = 11.949 (5) \text{ \AA}$

$\beta = 121.110 (14)^\circ$

$V = 1188.0 (8) \text{ \AA}^3$

$Z = 4$

$F_{000} = 528$

$D_x = 1.416 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

$\lambda = 0.71070 \text{ \AA}$

Cell parameters from 5144 reflections

$\theta = 3.4\text{--}29.6^\circ$

$\mu = 0.26 \text{ mm}^{-1}$

$T = 98 (2) \text{ K}$

Prism, colourless

$0.42 \times 0.12 \times 0.02 \text{ mm}$

Data collection

Rigaku AFC12κ/SATURN724
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 98(2) \text{ K}$

ω scans

Absorption correction: multi-scan
(ABSCOR; Higashi, 1995)

$T_{\min} = 0.506, T_{\max} = 1.000$

2202 independent reflections

2160 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.023$

$\theta_{\text{max}} = 26.5^\circ$

$\theta_{\text{min}} = 2.0^\circ$

$h = -25 \rightarrow 25$

$k = -7 \rightarrow 6$

6334 measured reflections

$l = -14 \rightarrow 14$

Refinement

Refinement on F^2

Hydrogen site location: inferred from neighbouring sites

Least-squares matrix: full

H-atom parameters constrained

$R[F^2 > 2\sigma(F^2)] = 0.030$

$$w = 1/[\sigma^2(F_o^2) + (0.0306P)^2 + 1.0876P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$wR(F^2) = 0.069$

$(\Delta/\sigma)_{\max} = 0.001$

$S = 1.05$

$\Delta\rho_{\max} = 0.22 \text{ e } \text{\AA}^{-3}$

2202 reflections

$\Delta\rho_{\min} = -0.19 \text{ e } \text{\AA}^{-3}$

163 parameters

Extinction correction: none

1 restraint

Absolute structure: Flack (1983), 845 Friedel pairs

Primary atom site location: structure-invariant direct methods

Flack parameter: 0.03 (7)

Secondary atom site location: difference Fourier map

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.53512 (3)	0.82070 (8)	0.39615 (4)	0.02211 (12)
N1	0.45431 (9)	1.0153 (3)	0.15952 (14)	0.0166 (3)
N2	0.43439 (9)	1.0280 (3)	0.02939 (14)	0.0192 (3)
N3	0.55419 (8)	0.7817 (3)	0.19105 (13)	0.0170 (3)
H3	0.5407	0.8284	0.1118	0.020*
C1	0.38502 (11)	1.1986 (4)	-0.02293 (18)	0.0205 (4)
H1	0.3618	1.2443	-0.1119	0.025*
C2	0.36965 (9)	1.3095 (4)	0.06872 (15)	0.0183 (4)
C3	0.32405 (11)	1.4985 (4)	0.0617 (2)	0.0230 (4)
H3A	0.2935	1.5815	-0.0178	0.028*
C4	0.32458 (11)	1.5613 (4)	0.1736 (2)	0.0244 (4)
H4	0.2943	1.6899	0.1717	0.029*
C5	0.36970 (11)	1.4360 (4)	0.29058 (19)	0.0237 (4)
H5	0.3689	1.4827	0.3662	0.028*
C6	0.41493 (11)	1.2485 (4)	0.29997 (17)	0.0215 (4)
H6	0.4448	1.1652	0.3796	0.026*
C7	0.41481 (10)	1.1861 (3)	0.18606 (17)	0.0167 (4)
C8	0.51513 (10)	0.8685 (3)	0.24509 (16)	0.0174 (4)
C9	0.61595 (10)	0.6190 (3)	0.25473 (16)	0.0154 (4)
C10	0.60810 (11)	0.4125 (4)	0.30926 (17)	0.0189 (4)

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H10	0.5610	0.3779	0.3039	0.023*
C11	0.66871 (11)	0.2582 (3)	0.37105 (17)	0.0205 (4)
H11	0.6635	0.1188	0.4095	0.025*
C12	0.73743 (10)	0.3055 (4)	0.37739 (16)	0.0208 (4)
H12	0.7793	0.1996	0.4207	0.025*
C13	0.74437 (11)	0.5080 (3)	0.32012 (18)	0.0201 (4)
H13	0.7908	0.5393	0.3226	0.024*
C14	0.68392 (10)	0.6660 (3)	0.25904 (17)	0.0185 (4)
H14	0.6891	0.8054	0.2205	0.022*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0249 (2)	0.0273 (3)	0.0173 (2)	0.0050 (2)	0.01318 (17)	0.0041 (2)
N1	0.0174 (8)	0.0192 (8)	0.0148 (7)	0.0017 (6)	0.0094 (6)	0.0010 (6)
N2	0.0223 (8)	0.0217 (9)	0.0152 (7)	0.0009 (7)	0.0109 (6)	0.0016 (6)
N3	0.0174 (7)	0.0209 (9)	0.0143 (6)	0.0022 (6)	0.0093 (6)	0.0031 (6)
C1	0.0216 (9)	0.0223 (10)	0.0186 (9)	0.0037 (8)	0.0111 (7)	0.0034 (8)
C2	0.0162 (8)	0.0204 (9)	0.0195 (8)	-0.0004 (9)	0.0101 (7)	-0.0003 (9)
C3	0.0210 (10)	0.0219 (10)	0.0248 (10)	0.0021 (8)	0.0110 (8)	0.0013 (8)
C4	0.0196 (10)	0.0234 (11)	0.0323 (10)	0.0033 (8)	0.0149 (8)	-0.0027 (8)
C5	0.0217 (10)	0.0288 (11)	0.0236 (10)	-0.0023 (9)	0.0139 (8)	-0.0074 (8)
C6	0.0201 (9)	0.0276 (12)	0.0203 (9)	-0.0011 (8)	0.0129 (8)	-0.0024 (8)
C7	0.0145 (9)	0.0161 (9)	0.0213 (9)	-0.0019 (7)	0.0105 (7)	-0.0022 (7)
C8	0.0162 (8)	0.0189 (11)	0.0170 (8)	-0.0020 (7)	0.0086 (7)	-0.0004 (7)
C9	0.0168 (9)	0.0154 (9)	0.0116 (8)	0.0008 (7)	0.0056 (7)	-0.0005 (7)
C10	0.0191 (9)	0.0203 (9)	0.0168 (8)	-0.0011 (8)	0.0091 (7)	-0.0006 (7)
C11	0.0240 (10)	0.0197 (11)	0.0168 (8)	-0.0001 (8)	0.0097 (7)	-0.0007 (7)
C12	0.0175 (8)	0.0219 (9)	0.0180 (8)	0.0024 (9)	0.0055 (7)	-0.0035 (8)
C13	0.0182 (9)	0.0213 (10)	0.0219 (9)	-0.0016 (8)	0.0112 (8)	-0.0058 (8)
C14	0.0199 (9)	0.0181 (10)	0.0182 (9)	-0.0020 (7)	0.0104 (7)	-0.0018 (7)

Geometric parameters (\AA , $^\circ$)

S1—C8	1.6545 (19)	C5—C6	1.377 (3)
N1—N2	1.391 (2)	C5—H5	0.9500
N1—C7	1.399 (2)	C6—C7	1.405 (3)
N1—C8	1.405 (2)	C6—H6	0.9500
N2—C1	1.303 (3)	C9—C14	1.387 (3)
N3—C8	1.352 (2)	C9—C10	1.393 (3)
N3—C9	1.426 (2)	C10—C11	1.379 (3)
N3—H3	0.8800	C10—H10	0.9500
C1—C2	1.432 (3)	C11—C12	1.391 (3)
C1—H1	0.9500	C11—H11	0.9500
C2—C3	1.396 (3)	C12—C13	1.384 (3)
C2—C7	1.405 (3)	C12—H12	0.9500
C3—C4	1.378 (3)	C13—C14	1.390 (3)
C3—H3A	0.9500	C13—H13	0.9500
C4—C5	1.407 (3)	C14—H14	0.9500

C4—H4	0.9500		
N2—N1—C7	110.34 (14)	N1—C7—C2	106.18 (15)
N2—N1—C8	118.84 (14)	N1—C7—C6	132.99 (17)
C7—N1—C8	130.08 (15)	C2—C7—C6	120.82 (18)
C1—N2—N1	106.36 (15)	N3—C8—N1	112.77 (15)
C8—N3—C9	123.98 (14)	N3—C8—S1	125.95 (14)
C8—N3—H3	118.0	N1—C8—S1	121.26 (13)
C9—N3—H3	118.0	C14—C9—C10	120.00 (17)
N2—C1—C2	112.47 (17)	C14—C9—N3	119.07 (16)
N2—C1—H1	123.8	C10—C9—N3	120.92 (17)
C2—C1—H1	123.8	C11—C10—C9	119.97 (18)
C3—C2—C7	121.21 (16)	C11—C10—H10	120.0
C3—C2—C1	134.13 (17)	C9—C10—H10	120.0
C7—C2—C1	104.65 (17)	C10—C11—C12	120.40 (18)
C4—C3—C2	118.07 (18)	C10—C11—H11	119.8
C4—C3—H3A	121.0	C12—C11—H11	119.8
C2—C3—H3A	121.0	C13—C12—C11	119.46 (18)
C3—C4—C5	120.36 (19)	C13—C12—H12	120.3
C3—C4—H4	119.8	C11—C12—H12	120.3
C5—C4—H4	119.8	C12—C13—C14	120.55 (17)
C6—C5—C4	122.72 (18)	C12—C13—H13	119.7
C6—C5—H5	118.6	C14—C13—H13	119.7
C4—C5—H5	118.6	C9—C14—C13	119.60 (18)
C5—C6—C7	116.82 (17)	C9—C14—H14	120.2
C5—C6—H6	121.6	C13—C14—H14	120.2
C7—C6—H6	121.6		
C7—N1—N2—C1	0.2 (2)	C5—C6—C7—N1	177.58 (19)
C8—N1—N2—C1	-170.97 (16)	C5—C6—C7—C2	-0.7 (3)
N1—N2—C1—C2	0.3 (2)	C9—N3—C8—N1	-176.69 (16)
N2—C1—C2—C3	178.1 (2)	C9—N3—C8—S1	5.0 (3)
N2—C1—C2—C7	-0.7 (2)	N2—N1—C8—N3	9.0 (2)
C7—C2—C3—C4	0.1 (3)	C7—N1—C8—N3	-160.10 (18)
C1—C2—C3—C4	-178.5 (2)	N2—N1—C8—S1	-172.60 (13)
C2—C3—C4—C5	-0.4 (3)	C7—N1—C8—S1	18.3 (3)
C3—C4—C5—C6	0.2 (3)	C8—N3—C9—C14	-129.93 (19)
C4—C5—C6—C7	0.4 (3)	C8—N3—C9—C10	51.7 (2)
N2—N1—C7—C2	-0.61 (19)	C14—C9—C10—C11	2.0 (3)
C8—N1—C7—C2	169.24 (17)	N3—C9—C10—C11	-179.65 (16)
N2—N1—C7—C6	-179.06 (19)	C9—C10—C11—C12	-1.1 (3)
C8—N1—C7—C6	-9.2 (3)	C10—C11—C12—C13	-0.6 (3)
C3—C2—C7—N1	-178.21 (17)	C11—C12—C13—C14	1.4 (3)
C1—C2—C7—N1	0.76 (19)	C10—C9—C14—C13	-1.2 (3)
C3—C2—C7—C6	0.5 (3)	N3—C9—C14—C13	-179.58 (16)
C1—C2—C7—C6	179.44 (17)	C12—C13—C14—C9	-0.5 (3)

Hydrogen-bond geometry (Å, °)

D—H...A	D—H	H...A	D...A	D—H...A
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supplementary materials

N3—H3…N2 ⁱ	0.88	2.30	3.097 (3)	151
C5—H5…S1 ⁱⁱ	0.95	3.14	3.881 (2)	136
C6—H6…S1 ⁱⁱⁱ	0.95	3.17	4.040 (3)	153
C1—H1…Cg(C9–C14) ^{iv}	0.95	???	2.47	???

Symmetry codes: (i) $-x+1, y, -z$; (ii) $-x+1, y+1, -z+1$; (iii) $-x+1, y, -z+1$; (iv) $-x+1, y+1, -z$.

Fig. 1

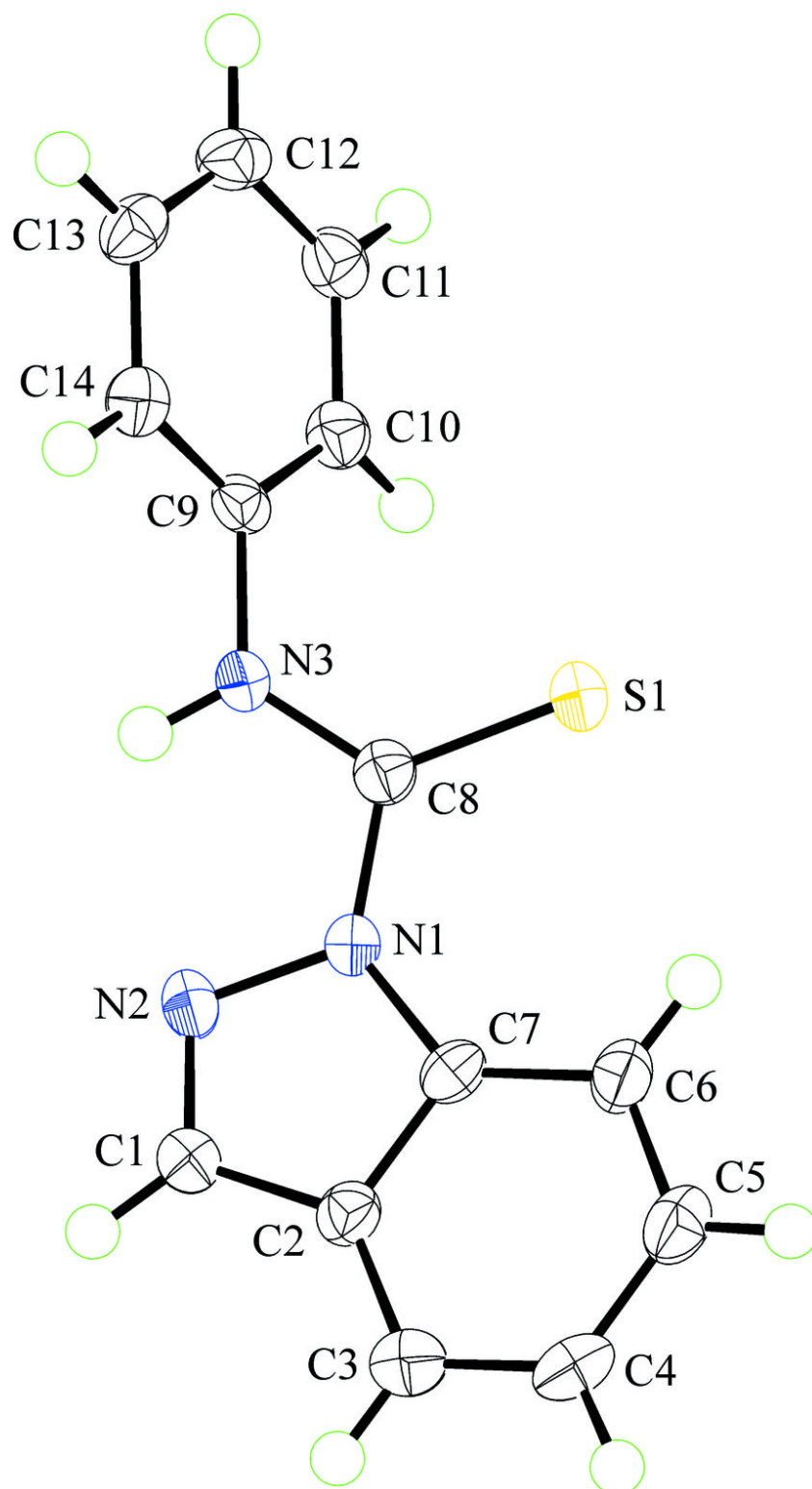


Fig. 2

